VIDYANIKETAN COACHING CLASSES, GHANSAANGI

Class:- 10th Sub.:- Science-1 Mark's:- 25 Time:- 1:30 Hr

[5]

Q.1) Choose the correct alternatives. [3] I] -----is a basic oxide. $[A] CO_2$ [B] K_2 0 $[C] SO_2$ $[D] Al_2 O_3$ ii] Bronze is an alloy of------[A] copper and tin [B] copper and zinc [C] copper and iron [D] iron and nickel Iii] -----is obtained by the reduction of its oxide by carbon. [A] Zinc [B] Aluminium [C] Sodium [D] Potassium Q.2) Answer the following question's. [Any-4] [8] I] Explain the term corrosion with suitable examples. ii] Why sodium is stored in kerosene? iii] Write names:- a.] Alloy of sodium with mercury. b.] The reagent that dissolves noble metals. Iv] Write chemical equation for the following events. a] Iron filings are dropped in aqueous solution of copper sulphate. b] Aluminium came in contact with air. v] Distinguish between Metal and Non-Metal. vi] Write the electrode reaction for electrolysis of molten magnesium chloride and calcium chloride. Q.3) Answer the following question's. [Any-3] [9] i] State the general properties of ionic compounds.

- ii] The electronic configuration of metal 'A' is 2,8,1 and that of metal 'B' is 2,8,2. Which of the two metals is more reactive? Write their reaction with dilute hydrochloric acid.
- iii] Explain the term:- a.] Metallurgy b.] Ores c.] Minerals
- iv] What are alloys? Explain with suitable example properties of alloy which makes them useful over pure metal.
- V] Describe the magnetic separation method?

Q.4) Answer the following.[Any-1]

- i] A scientist was studying the metals A, B and C. He found that A reacts with cold water, hot water and steam, B with hot water and steam. He found that C reacts with steam only. Identify A, B and C and write the chemical equations involved.
- ii] Describe the electrolytic reduction of alumina with neat labelled diagram.

Best of luck......

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